



UNIVERSITÀ
DEGLI STUDI DI MILANO-BICOCCA

SYLLABUS DEL CORSO

Chimica e Propedeutica Biochimica

1819-1-H4101D252-H4101D002M

Aims

The student must achieve the knowledge on: Atomic structure, Chemical bonds, Solutions. Chemical reactions and catalysis. Energy aspects of chemical reactions. Acids and bases and buffer solutions. Organic compounds and functional groups: structural properties of organic molecules and chemical reactivity. The compounds of life: carbohydrates, amino acids, nucleotides, lipids. Proteins. Polysaccharides. Nucleic acids.

Contents

The primary goal of the course is to provide the tools for the understanding of the complex reactions that represent the molecular basis of life and to give to the student the basis to identify the cause-effect relations of the most important chemical processes for the curriculum and the work of a physician. This knowledge will form the primary basis for a rationale approach to the knowledge of medical sciences.

Detailed program

BALANCE REACTIONS - Chemical equilibrium - Position of equilibrium, equilibrium constant and factors influencing it - The kinetics of chemical reactions - Speed of reaction and factors influencing it CATALYSIS OF CHEMICAL TRANSFORMATIONS - catalysts: how they modify the speed of chemical reactions - Enzyme catalysts and their activity and specificity

ENERGY ASPECTS OF CHEMICAL REACTIONS - Elements of electrochemistry: potential redox and spontaneity of redox reactions - Thermodynamic elements: state of a system, status functions and thermodynamic variables, thermodynamic transformations. - Thermodynamic elements: interpretation of the spontaneity of chemical reactions through the functions of state (entropy, enthalpy, free energy) - Spontaneity of equilibrium reactions.

ACIDS AND BASES - Acid-base balances: acid and base definitions, strength of acids and bases in water; approximate pH calculation - Acidity and basicity of salt solutions. Solution buffer and their buffering power - Blood buffer systems.

CLASSIFICATION, GENERAL PROPERTIES OF ORGANIC COMPOUNDS FINALIZED TO THE INTERPRETATION OF BIOCHEMICAL PROCESSES - Physico-chemical properties - Structural isomerism and stereoisomerism - Optical isomers and geometric isomers - Structure of coordination compounds and their biological importance

STRUCTURE, NOMENCLATURE AND REACTIVITY OF ORGANIC COMPOUNDS - Hydrocarbons, alcohols, thioalcohols and analogues; Amines, Carbonyl compounds Carboxylic acids and carboxylic acid derivatives, Polyfunctional compounds

COMPOUNDS OF RELEVANT BIOLOGICAL INTEREST - Lipids, Carbohydrates, Amino Acids, Nucleotides

BIOLOGICAL POLYMERS Polysaccharides, Peptides and Proteins, Nucleic Acids

Prerequisites

In order to standardize the basic knowledge of the class, the School of Medicine organizes lectures and tutorials in physics and chemistry preliminary to the course

Teaching form

Lectures, exercises, laboratories

It is required 70% course attendance

Textbook and teaching resource

A. Fiecchi, M. Galli Kienle, A. Scala Chimica e Propedeutica Biochimica Ed. Edi Ermes.

E. Santaniello, M. Alberghina, M. Coletta, S. Marini Principi di Chimica Generale e Organica Ed. PICCIN

F.A. Bettelheim, W.H. Brown, M.K. Campbell, S.O. Farrell Chimica e Propedeutica Biochimica EdiSES

Semester

First semester

Assessment method

No ongoing tests

The evaluation will consist of a written test that will be used to ascertain the level of knowledge and ability to understand the topics covered during the course and to be able to solve problems.

For these modules the student will have to answer:

CHEMISTRY AND PROPAEDEUTIC BIOCHEMISTRY and PROTEOMICS

4 Open Response Questions (also with numerical exercises) concerning the topics of general chemistry, organic chemistry, biological compounds and proteomics, respectively

15 Single-answer quiz with 5 answers, of which only one is correct

Through this written test, the expositive and synthesis skills will be evaluated.

Oral examination on the evaluation of the teachers (discussion of the written test) The oral test will serve to clarify critical issues emerged from the written test and to verify the communication skills of the student and will focus on the topics covered by the written test

Office hours
