UNIVERSITÀ
DEGLI STUDI DI MILANO-BICOCCA

SYLLABUS DEL CORSO

Struttura della Materia

2021-3-E3001Q057

Aims

Understanding the properties of atoms, molecules and solids by means of quantum mechanics and statistical mechanics.

Contents

Elements of classical and quantum statistical mechanics.

Atoms: two-electrons atoms, many-electron atoms in the Hartree theory and the periodic table.

Molecules: electronic states and the chemical bonding of diatomic and polyatomic molecules, molecular rotations and vibrations, molecular spectroscopy.

Solids: band theory of electrons in crystals, electrical conduction in metals, semiconductors and semiconductor devices.

Detailed program

The course consists of four sections after an introductory lecture on many particle systems.

The reference to specific chapters of the textbooks is given in each section.

Quantum Mechanics of Many Particles Systems

(**(G)**, chapter 13)

Identical particles: Fermions and Bosons, Slater determinant for independent particles, Pauli exclusion principle.

Statistical Physics

(**(KK)** chapters 2, 3, 6-9 or equivalently **(T)** chapters 1, 2.1-2.4, 3.4-3.5.3, 3.6.1-3.6.3 or **(M)** chapter 4)

- Entropy, temperature and probability.
- Canonical ensemble and the Boltzmann distribution.
- Ideal classical gas.
- Chemical potential, gran canonical ensemble and the Gibbs distribution.
- Quantum distributions functions: Fermi-Dirac and Bose-Einstein distributions.
- Degenerate Fermi gas: Fermi energy, specific heat.
- Low temperature Bose gas and the Bose-Einstein condensation, superfluid Helium.

Atomic physics

(**(G)** chapter 14 with supplements 14-A and 14-B, **(BJ)** chapters 7 and 8)

- Two-electrons atoms: perturbation theory and variational principle for the ground state.
- Exited states of two-electrons atoms: parahelium and orthoelium.
- Many-electron atoms in the Hartree theory.
- Ground state of many-electron atoms and the periodic system of the elements.
- Corrections to the central field approximation: L-S and j-j couplings, Hund's rules.

Molecular Physics

(**(M)** chapter 3, **(BJ)** chapters 10 and 11)

- The Born-Oppheneimer approximation.
- The electronic structure of the H_2 molecule: the Heitler-London and the molecular orbital schemes.
- Electronic states in homo- and hetero-nuclear diatomic molecules, covalent and ionic bonding.
- Electronic states in polyatomic molecules: hybridization and the Hueckel model.
- Rotations and vibrations of diatomic molecules.
- Raman and IR spectra of the diatomic molecule. IR selection rules in the electric dipole approximation.
- The effects of the nuclear spin on the rotation of the homonuclear diatomic molecules.
- Specific heat of polyatomic molecules. The theorem of equipartition of energy.

Solid State Physics

(**(M)** chapter 5)

- Lattices and crystal structures.
- Diffraction experiments and the reciprocal lattice.
- The band theory of electrons in crystals: metals and insulators.
- Semiclassical dynamics of electrons in crystals and the electrical conductivity of metals.
- Semiconductors: distribution of electrons and holes in intrinsic semiconductors, n and p doping, acceptor and donor states in the hydrogenic model.
- Semiconductor devices: the pn junction.
- The laser. (**(M)**, section 4.4.1)

Prerequisites

Mathematics and physics courses of the first two years. The first part of the course of Quantum Mechanics.

Teaching form

Frontal lessons.

In the period of Covid-19 emergency the lectures will be given remotely partially by videorecording in an synchronous manner and partially in an asynchronous manner. Periodic meetings in synchronous videoconference

will be planned.

Textbook and teaching resource

- S. Gasiorowicz, *Quantum Physics*, (Wiley International Editions, 2003) **(G)**

- C. Kittel e H. Kroemer, *Thermal Physics* (W. Freeman, 1980) or the Italian edition, _____

- N. Manini, *Introduction to the Physics of Matter*, (Springer, 2014), available in e-book on the library website. **(M)**

- B. H. Bransden & C. J. Joachain, *Physics of Atoms and Molecules,* 2nd edition, (Harlow – Prentice Hall, 2003). **(BJ)**

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for further details on special topics

C. Kittel, Introduction to Solid State Physics, 8th edition, Wiley (2005) or the italian edition published by Editrice Ambrosiana.

H. Haken and H. C. Wolf, Molecular Physics and Elements of Quantum Chemistry, Springer, available in e-book on the library website. Cap. 4, 5, 9-13.

Semester

First and second semester.

Assessment method

Students are evaluated through a written exam followed by an oral one.

The written exam lasting 2.5 hours consists of four numerical exercises on topics of statistical mechanics, atomic and molecular physics and solid state physics.

A mark of 7.5 points will be given for each exercise correctly solved.

To be admitted to the oral exam a minimum mark of 11 points is required (1.5 exercise correctly solved).

The oral exam will be focused on the discussion of the written exam and on the topics of the lessons.

The oral exam must be scheduled in the same session of the written exam.

No ongoing partial tests are planned.

In the period of Covid-19 emergency the written exam will take place remotely by using the webex platform. The written exam will last 2 hours for the solution of three exercises. To be admitted to the oral exam a minimum mark of 7.5 points is required (1 exercise correctly solved out of three).

Instructions for the written exam via webex are available on the e-learning webpage of the course as a pdf file.

A webex link will be available on the e-learning webpage for the access to the written exam restricted to students registered for the exam.

The oral exams will be in video conference as well via the WebEx platform to be accessed from the e-learning webpage. For oral exams a public webex link will be available to be distributed to anyone willing to attend at the exams.

Office hours

At 17:30 pm on the days on which a lecture ending at 17:30 is scheduled.

By appointment in the periods with no lessons.