UNIVERSITÀ
DEGLI STUDI DI MILANO-BICOCCA

COURSE SYLLABUS

Chemistry and Introduction To Biochemistry

2122-1-H4601D069-H4601D002M

Aims

The student must achieve the knowledge on: Atomic structure, Chemical bonds, Solutions. Chemical reactions and catalysis. Energy aspects of chemical reactions. Acids and basis and buffer solutions. Organic compounds and functional groups: structural properties of organic molecules and chemical reactivity. The compounds of life: carbohydrates, amino acids, nucleotides, lipids. Proteins. Polysaccharides. Nucleic acids.

Contents

The primary goal of the course is to provide the tools for the understanding of the complex reactions that represent the molecular basis of life and to give to the student the basis to identify the cause-effect relations of the most important chemical processes for the curriculum and the work of a physician. This knowledge will form the primary basis for a rational approach to the knowledge of medical sciences.

Detailed program

BALANCE REACTIONS - Chemical equilibrium - Position of equilibrium, equilibrium constant and factors influencing it - The kinetics of chemical reactions - Speed of reaction and factors influencing it CATALYSIS OF CHEMICAL TRANSFORMATIONS - catalysts: how they modify the speed of chemical reactions - Enzyme catalysts and their activity and specificity

ENERGY ASPECTS OF CHEMICAL REACTIONS - Elements of electrochemistry: potential redox and spontaneity of redox reactions - Thermodynamic elements: state of a system, status functions and thermodynamic variables,

thermodynamic transformations. - Thermodynamic elements: interpretation of the spontaneity of chemical reactions through the functions of state (entropy, enthalpy, free energy) -Spontaneity of equilibrium reactions.

ACIDS AND BASES - Acid-base balances: acid and base definitions, strength of acids and bases in water; approximate pH calculation - Acidity and basicity of salt solutions. Solution buffer and their buffering power - Blood buffer systems.

CLASSIFICATION, GENERAL PROPERTIES OF ORGANIC COMPOUNDS FINALIZED TO THE INTERPRETATION OF BIOCHEMICAL PROCESSES - Physico-chemical properties - Structural isomerism and stereoisomerism - Optical isomers and geometric isomers - Structure of coordination compounds and their biological importance

STRUCTURE, NOMENCLATURE AND REACTIVITY OF ORGANIC COMPOUNDS - Hydrocarbons, alcohols, thioalcohols and analogues; Amines, Carbonyl compounds Carboxylic acids and carboxylic acid derivatives, Polyfunctional compounds

COMPOUNDS OF RELEVANT BIOLOGICAL INTEREST - Lipids, Carbohydrates, Amino Acids, Nucleotides

BIOLOGICAL POLYMERS Polysaccharides, Peptides and Proteins, Nucleic Acids

Prerequisites

none

Teaching form

Lectures, exercises, laboratories

It is required 70% course attendance

Lesson in attendance, subject to any ministerial changes following the COVID pandemic situation

Textbook and teaching resource

F.A. Bettelheim, W.H.Brown, M.K. Campbell, S.O. Farrell Chimica e Propedeutica Biochimica EdiSES

E. Santaniello, M. Alberghina, M. Coletta, S. Marini Principi di Chimica Generale e Organica Ed.PICCIN

A. Fiecchi, M. Galli Kienle, A. Scala Chimica e Propedeutica Biochimica Ed. Edi Ermes.

Semester

First Semester

Assessment method

The evaluation will consist of a written test that will be used to ascertain the level of knowledge and ability to understand the topics covered during the course and to be able to solve problems.

For these modules the student will have to answer:

3 Open Response Questions (also with numerical exercises) concerning the topics of general chemistry, organic chemistry, biological compounds respectively

10 Single-answer quiz with 5 answers, of which only one is correct

In case of Covid-19 emergency, exams will be online. They will be carried out using the Proctoring/Respoundus system.

A public link for access to the examination of possible virtual spectators will be reported on the e-learning page of the course.

Office hours

By appointment (e-mail) at Building U28, I floor