

COURSE SYLLABUS

Advanced Physical Chemistry - Module 2

2324-1-F5401Q027-M2

Aims

Aim of this class is to provide students with an introduction to statistical mechanics with its chemical applications

Knowledge and understanding - Methods of statistical analysis of many-particle systems

- Procedures for the calculation of thermodynamic quantities from microscopic models of many-particle systems
- Basic mathematical tools for the calculation of the partition function
- Methods suitable for the approximate description of classical and quantum interacting systems

Applying knowledge and understanding

- Computation of the equations of state and of the thermodynamic potentials of chemical systems
- Evaluation of the limit of applicability of the classical approximation in chemistry

Making judgements - Evaluation of the appropriateness of numerical calculation techniques used in commercial simulation codes

- Capabilities of critical analysis of axiomatic scientific theories

Communication skills

Rigorous use of natural language in science

Learning skills Activation of critical skills in the analysis of scientific models

Contents

Ensembles and phase space. The density of states. The principle of equal a priori probability. Criteria of statistical equilibrium. The Liouville and H theorems. Micro-canonical, canonical and grand-canonical ensembles. Classical and quantum perfect gas.

Detailed program

The equation of motion in the Lagrangian form. Generalised momenta and the canonical equation. Statistical ensembles and the phase space. The density of states. The principle of equal a priori probability. The Liouville theorem. Criteria of statistical equilibrium. Micro-canonical, canonical and grand-canonical ensembles. The Maxwell-Boltzmann distribution for a micro-canonical ensemble. The principle of equipartition. The Boltzmann's H-theorem. Applications: free particles, particles in a box, particles in a harmonic force field, particles with spin. Applications to thermodynamic systems relevant to chemistry: the monoatomic ideal gas, ideal gas mixtures, non-ideal gases. An outline about Bose-Einstein and Fermi-Dirac quantum distributions.

Prerequisites

Classical thermodynamics, calculus of functions of several variables, fundamentals of quantum mechanics.

Teaching form

The class is made of two sub-units, delivered by Claudio Greco (quantum mechanics) and by Dario Narducci (statistical mechanics), by classroom lectures.

The sub-unit of Statistical Mechanics will be delivered through classroom lectures.

Lectures will be in Italian in the absence of Erasmus students; in English otherwise.

Textbook and teaching resource

Reif, Frederick, *Fundamentals of statistical and thermal physics*, McGraw-Hill, 1965 e Waveland Press, 2009

Semester

First year, first term

Assessment method

Oral exam. Students may opt for partial oral tests, one for each sub-unit. No mid-term tests.

The oral exam of Statistical Mechanics aims at verifying the level of knowledge acquired, the understanding of the main conceptual junctures in the development of the theory presented during the class and the appropriate use of the language by the student.

Office hours

By appointment.

Sustainable Development Goals
