



UNIVERSITÀ
DEGLI STUDI DI MILANO-BICOCCA

COURSE SYLLABUS

General Chemistry Laboratory

2526-3-E1301Q084-E1301Q081M

Aims

The Organic Chemistry module allows the student to become confident with analytical and preparative chromatographic techniques and with reactivity of organic compounds applied to biological systems.

1. Knowledge and understanding.

At the end of the course, students will know basic theory and experimental of chromatography and chemical transformations of organic compounds.

2. Ability to apply knowledge and understanding.

At the end of the course, students will be able to apply the acquired knowledge to organic compounds transformation and purification.

3. Making judgements.

At the end of the course, students will be able to process what they have learned to general experimental methodologies.

4. Communication skills.

At the end of the course, students will be able to process the experimental data obtained and to describe the procedures and the results, using the most appropriate technical vocabulary.

5. Learning skills.

At the end of the course, students will be able to apply basic experimental techniques of the organic chemistry lab to biomolecules.

Contents

The experiences of general chemistry laboratory consist of experiments in the field of determination of solute

concentration in diluted aqueous solutions using basic techniques of quantitative analytical chemistry.

Detailed program

The general chemistry module will be organised in 5 experimental lab sessions, focussed on the following techniques and methodologies:

Acid-base titration of hydrochloric acid with sodium carbonate with methyl orange indicator; Redox titration of hydrogen peroxide with a solution of potassium permanganate standardized with sodium oxalate; Iodometric titration of commercial sodium hypochlorite (bleach) with sodium thiosulfate; Acid-base pHmetric titration of phosphoric acid in an unknown solution and in coca-cola with sodium hydroxide; Determination of the isoelectric point of glycine; Colorimetric analysis of a diluted solution of Fe^{2+} .

Prerequisites

Basic general chemistry and stoichiometry

Teaching form

Laboratory experimental activities conducted in equipped laboratories.

Four laboratory activities, 5 hours each, consisting of 1 hour of delivered teaching and 4 hours of Interactive Teaching. Attendance is mandatory

Teaching language: italian.

Textbook and teaching resource

Slides and experimental protocols, illustrative videos and self-assessment tests will be made available on the e-learning platform.

Semester

Second semester

Assessment method

For the general Chemistry module, the assessment method is a written test of 1 hour, to be held in the computer room, through the e-learning platform, and aimed at assessing the acquired skills .

Office hours

The teachers will receive by appointment requested by e-mail.

Sustainable Development Goals

GOOD HEALTH AND WELL-BEING
