



UNIVERSITÀ
DEGLI STUDI DI MILANO-BICOCCA

COURSE SYLLABUS

Biological Chemistry

2526-2-E1301Q073

Aims

The course imparts basic knowledge and concepts of biochemistry, which represent the prerequisite to address any further biological subject at the molecular and cellular level. In particular, it deals with protein chemistry, enzymology and metabolism. In the framework of these topics, only fundamental contents are presented, without a systematic treatment of biochemistry. In fact, the course is mainly aimed at clarifying the rationale underlying the described phenomena, thus providing a key to the understanding of molecular and metabolic processes as a whole.

A significant part of the teaching will be devoted to theoretical exercises, whereby students will be familiarized with basic chemical-physical laws that govern biological phenomena.

1. Knowledge and understanding. Knowledge imparted provides the basic tools for the understanding of biological phenomena at the molecular and cellular level.
2. Applying knowledge and understanding. Knowledge imparted represents the basic background to address experimental issues in biology.
3. Making judgments. The course provides a knowledge basis for the understanding of scientific literature and for the critical assessment of issue related to biological experimentation.
4. Communication skills. The course imparts basic elements of scientific language currently adopted in the context of biological research.
5. Learning skills. Knowledge provided by the course represents a prerequisite for effective learning of contents subsequently imparted in several areas of biology.

Contents

FIRST MODULE:

- Non-covalent interactions in aqueous solution
- Structural levels and properties of proteins

- Enzymology
- Allosteric proteins

SECOND MODULE:

- Metabolism

Detailed program

FIRST MODULE:

1 - GENERAL PRINCIPLES

Structure and general properties of water. Intra- and intermolecular non-covalent interactions: their role in determining the properties of biological molecules. Acid-base dissociation in aqueous solutions. Buffer solutions.

2 – LEVELS OF STRUCTURAL ORGANIZATION IN PROTEINS

Structures of amino acids found in proteins. Physical and acid-base properties of amino acids. Properties of the peptide bond. Filamentous proteins. The different structural levels in proteins. Physical properties of proteins. Criteria of classification of proteins. Basic principles of protein folding. Methods of protein 3D structure prediction.

3 - ENZYMES

Mechanisms of enzyme catalysis. Examples of mechanisms of enzyme-catalyzed reactions. Steady-state enzyme kinetics. Physical factors affecting enzyme activity. Regulation of the enzyme activity: the role of inhibitors and activators.

4 – ALLOSTERIC PROTEINS

Definition and adaptive roles of allosteric behavior. Molecular mechanisms of allostericity. Interpretation of allosteric behavior at the molecular level. Some examples of allosteric proteins and enzymes. The globins.

SECOND MODULE:

5 – METABOLISM

Basic principles. Glycolysis and fermentations. The Krebs cycle. The oxidative phosphorylation. The pentose-phosphate pathway. Biosynthesis and degradation of lipids. Biosynthesis and degradation of amino acids. Glycogen metabolism. Gluconeogenesis. The nitrogen cycle in the biosphere. Integration of metabolic pathways. Basic concepts on the role and the action mechanisms of hormones.

Prerequisites

General and organic chemistry. Basic knowledge of cell biology.

Teaching form

64 h lessons, 2 h each: a section of delivered didactics (Didattica erogativa, DE) focused on the presentation-illustration of contents by the lecturer; a section of interactive teaching (Didattica Interattiva, DI) including teaching interventions supplementary to delivered didactic activities and discussion of case studies.

Didactic activities are conveyed by means of face-to-face lectures

Teaching language: Italian

Textbook and teaching resource

- Campbell e Farrell: “Biochimica”, Edises
- Nelson, Cox: “Principi di biochimica di Lehninger”, Ed. Zanichelli
- Mathews, Van Holde, Ahern: “Biochimica”, Casa Editrice Ambrosiana.
- A text drawn up by the teacher is available online for free (e-learning).

Semester

1st semester.

Assessment method

Students must take two multiple choice tests (one for each teaching module) addressing all contents of the respective module. This is followed by a final oral examination, whereby students must prove to be acquainted with the basic concepts of the taught subjects.

Office hours

The teachers receive by appointment.

Sustainable Development Goals

QUALITY EDUCATION
